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3 \square ___ KCl + ___ O₂ a. 3 b. 5 c. 6 d. 7 e. 8 6. Calculate the mass of hydrogen formed when 27 g of aluminum reacts with excess hydrochloric acid according to the balanced equation below. $2\text{Al} + 6\text{HCl} \square 2 \text{AlCl}_3 + 3 \text{H}_2$ a. 1.5 g b. 2.0 g c. 3.0 g d. 6.0 g e. 12 g 7.

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a) Identify and describe the proton, neutron and electron in terms of their relative charges and relative masses. b) Discuss the behaviour of beams of protons, neutrons and electrons in electric fields. c)

Calculate the distribution of mass and charges within an atom from

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the given data. d) Deduce the number of protons, neutrons and electrons present in both atoms and ions for a given proton and nucleon numbers/charge.

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CBSE 10 Chemistry, Free Sample Papers, Sample Questions ... Chapter 3 Test Multiple Choice Identify the choice that best completes the statement or answers the question. 1. Most of the volume of an atom is occupied by the a. protons. c. nucleus. b. electron cloud. d. nuclides. 2. When the light from excited atoms of an element is passed through a prism, the distinct colored lines produced are called 3.

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statement or answers the question. 1) Silver has two naturally occurring isotopes with the following isotopic masses: 107.47 Ar 107.47 Ar 106.90509 108.9047 The average atomic mass of silver is 107.8682 amu .

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